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Mapping the Ultrafast Vibrational Dynamics of all-*trans* and 13-*Cis* Retinal Isomerization in Anabaena Sensory Rhodopsin

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Discrepancies in the isomerization dynamics and quantum yields of the *trans* and *cis* retinal protonated Schiff base is a well-known issue in the context of retinal photochemistry. Anabaena Sensory Rhodopsin (ASR) is a microbial retinal protein that comprises a retinal chromophore in two ground state (GS) conformations: all-*trans*, 15-*anti* (AT) and 13-*cis*, 15-*syn* (13C). In this work, we apply impulsive vibrational spectroscopic techniques (DFWM, pump-DFWM and pump-IVS) to ASR to shed more light on how the structural changes take place in the excited state within the same protein environment. Our findings point to distinct features in the ground state structural conformations as well as to drastically different evolutions in the excited state manifold. The ground state vibrational spectra show stronger Raman activity of the C₁₄-H out-of-plane wag (at about 805 cm⁻¹) for 13C than for AT isomer, which hints at a pre-distortion of the 13C in the ground state. Evolution of the Raman frequency after interaction with actinic pulse shows a blue shift for the C=C stretching and CH₃ rocking mode for both isomers. For AT, however, the blue shift is not instantaneous as observed for the 13C isomer, rather it takes more than 200 fs to reach the maximum frequency shift. This frequency blue shift is rationalized by a decrease of the effective conjugation length during the isomerization reaction, which further confirms a slower formation of the twisted state for the AT isomer and corroborates the presence of a barrier in the excited state trajectory previously predicted by quantum chemical calculations.

Introduction

Photo-induced isomerization of retinal protonated Schiff base (RPSB) powers many fundamental biological processes like photosynthesis, vision, gene expressions by initiating the photocycle via the conversion of light energy to chemical potential.¹⁻⁸ Retinal derivative, which serves as the central element for these photo-chemically triggered biological activity, is found to be covalently bound to a Lysine residue of the seventh helix of the protein membrane in all retinal pigments.^{9, 10} Retinal is well known for its different structural isomeric forms in a variety of organisms e.g. 11-*cis* in visual rhodopsin, 9-*cis* in iso-rhodopsin, all-*trans* in microbial retinal protein (MRP) and the widely varying reaction rates and the isomerization efficiencies in these proteins. For instance, the retinal isomerization in visual pigments takes place within 100 fs¹¹⁻¹⁴ with a quantum yield about 65%¹⁵ whereas it takes more than 0.6 ps to complete the isomerization with a yield

<40% for MRP's.¹⁶⁻¹⁸ On the other hand, the photo-isomerization of RPSB (all-*trans*) in solution happens in a much slower rate (~1 ps)¹⁹⁻²¹ and leads to the formation of a mixture of different stereo-isomers with a quantum yield of few percent for each subproduct.²² All these observations have been taken as an indication that the electrostatic interaction between the retinal chromophore and the amino acid residues of the surrounding protein steers the isomerization reaction.

In recent decades, there have been extensive experimental^{12, 14, 23-29} and theoretical³⁰⁻³⁴ investigations performed to explore the underlying mechanism of this ultrafast isomerization reaction. Visual rhodopsin and bacteriorhodopsin (BR) are two retinal proteins which have been most thoroughly studied in this regard. In general, the retinal isomerization is believed to be initiated by a rapid reorganization of C-C bond length within few tens of femtosecond after the excitation, known as bond length alternation (BLA),³² to form the sub-picosecond reactive excited state coined as I intermediate.³⁵ For BR, this state decays non-exponentially to form J intermediate^{35, 36} during the course of internal conversion (IC) which is associated with a weak spectral evolution.^{37, 38} Moreover, the cross section of the emission state has been found to be constant throughout the fluorescence lifetime.³⁸ Both of these observations have been interpreted as a non-ballistic internal conversion. Retinal isomerization in visual rhodopsin, however, differs from that: Dramatic spectral evolution during the IC^{12, 14, 27} has been interpreted as a coherent isomerization reaction^{12, 14, 39} where a

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nuclear wave packet generated by impulsive stimulated Raman scattering (ISRS) evolves in a ballistic fashion towards the ground state potential surface.

This contrast between the nature and speed of the isomerization reactions for two different classes of retinal proteins has been often rationalized by the difference in the planarity of the ground state (GS) structures of RPSB bound inside the cavity of retinal pocket.⁴⁰ Structural strains on RPSB inside different protein pockets, originate either due to distinct structural conformations of each isomer or due to the change in the electrostatic interaction between RPSB and the protein moiety. Structural investigations^{9, 40, 41} have shown that 11-*cis* GS isomer in visual rhodopsin is present as a non-planar, pre-twisted structure which facilitates the reaction to proceed rapidly.⁴² However, for BR neither NMR studies^{40, 43} nor the resonance Raman spectra⁴⁴ indicates any evidence of such a pre-twisting for the GS all-*trans* isomer. Furthermore, a quantum chemical computational study⁴² of an artificially twisted retinal in gas phase showed sub-100 fs dynamics which further indicates the pre-straining alone can lead to a rapid isomerization process even in absence of any specific electrostatic interaction of RPSB with the surrounding.

One systematic way to point out the reason would be to investigate RPSB of two different conformations under the same protein environment. Recently discovered Anabaena Sensory Rhodopsin (ASR) has been assisting to shed new light on this issue.^{7, 9, 45-47} Like other MRP's, it comprises 13-*cis*, 15-*syn* (13C) and all-*trans*, 15-*anti* (AT) in GS. Being a photochromic sensor, the isomeric ratio depends on the wavelength of external illumination light.⁴⁵ When it is illuminated with an orange light (~590 nm) it forms a mixture (~40:60) of AT and 13C isomers in light adapted (LA) photo-stationary equilibrium.^{46, 47} On the other hand, it adopts a thermally relaxed AT form under dark adapted (DA) condition. Within the photocycle (Fig. 1), each of the AT and 13C isomers undergoes isomerization around C₁₃=C₁₄ in sub-ps time scale that results in a hot photo-intermediate J (13-*cis*, 15-*anti* and all-*trans*, 15-*syn*, respectively). Later, it forms the K-photoproduct (K_{AT} and K_{13C}) within 100 ps via vibrational relaxation. Subsequently, both K_{AT} and K_{13C} undergo further isomerization around C₁₅=N on a longer time scale (<nS) to generate the 13C and AT GS respectively (Fig. 1). Therefore, ASR is a unique member among the rhodopsin family which allows to compare the isomerization reaction in both directions (AT to 13C and 13C to AT) of the photo-cycle within the same protein environment. This is extremely advantageous to evaluate the individual effect of GS conformation on the isomerization dynamics. Pump-probe spectroscopy has already showed the stark contrast in the reaction kinetics of AT and 13C isomer in ASR.^{46, 47} It has been observed that the 13C isomer shows a ballistic kinetics⁴⁶ and the isomerization completes within 100 fs, which is very similar to visual rhodopsin.^{46, 47} The AT isomer shows, however, about seven fold (about 750 fs)^{46, 47} slower kinetics, which is reminiscent to BR. This large disparity in the reaction dynamics has been qualitatively explained by the quantum chemical excited trajectory calculation, which suggests the existence of a small barrier or plateau in the

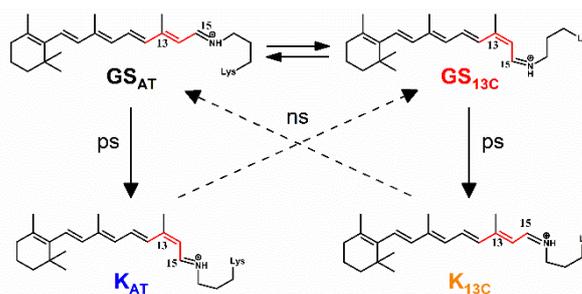


Figure 1: Schematic representation of the photocycle of AT and 13C isomer of ASR. After excitation, each isomer undergoes isomerization around C₁₃=C₁₄ bond in sub-ps timescale to form the corresponding hot photo-product, J_{AT} and J_{13C}, which generate K_{AT} and K_{13C} within 100 ps, respectively. Subsequent isomerization around C₁₅=N occurs on sub-ns time scale to complete the photo-cycle.

excited potential energy surface of AT isomer but not for the 13C isomer.³¹ The quantum yield (QY) of the photoproduct after the isomerization for each direction is also very different: it is about 2.7 times higher for the AT than that of the 13C isomer.⁴⁸ Lower quantum yields with short excited state lifetimes (ESL), often observed also in other proteins, has often been interpreted as a discrepancy lacking clarification. Ultimately, the ESL and QY are two fundamentally independent quantities: While the former depends on the topology near the Franck-Condon region and potential barriers in the ES surface, the latter depends on the shape of conical intersection (CI) seam and the wave packet trajectories. A potential way to address this persisting question is the comparison of isomer specific ultrafast structural changes in each case, which will give more insight into the molecular origin of this difference in potential energy surfaces.

In the context of retinal isomerization mechanism, a few vibrational modes such as torsion, C=C and C-C stretching, hydrogen out of plane (HOOP) wags are well known to participate in this reaction. A detailed study of the origin and evolution of the vibrational modes in the ground and excited state has been reported mainly for visual rhodopsin,^{23, 28, 49} BR^{24, 25, 50} and RPSB in solution.⁵¹⁻⁵⁴ Low frequency torsional modes (<400 cm⁻¹) are well accepted as one of the key features of ES wave packet dynamics. In particular, a set of rapidly damped low frequency coherences (100-400 cm⁻¹) has been observed in the spectral region of stimulated emission and excited state absorption.^{28, 50, 52-57} Pump-impulsive vibrational spectroscopic studies^{28, 53, 56} on RPSB both in solution and inside the protein environment have shown that these modes are activated only after the initial excited relaxation of RPSB. Beside the low frequency torsion, the HOOP wag (800-1000 cm⁻¹) is the most frequently discussed vibrational mode. This out of plane mode, being A_u/ A₂ symmetric, is Raman inactive for RPSB of planar geometry which belongs to C_{2v} (*cis*) or C_{2h} (all-*trans*) point group and gets Raman activated, only when the RPSB adopts a distorted non-planar structure. Hence, the activity of this mode is an indicator of non-planarity or pre-twisting of the RPSB in GS, as well as of the degree of distortion during the isomerization due to the rotation around the isomerizing C=C bond.^{23, 28, 58} Moreover, a number of experimental observations^{28, 59} and quantum chemical computations^{32, 60} have also suggested that

the phase and amplitude of the HOOP mode control the stereochemical outcome of the retinal photo-isomerization. Finally, high frequency ($>1000\text{ cm}^{-1}$) wave packet dynamics have been very often interpreted as GS wave packet motion^{25, 61} with very few reports^{62, 63} proposed that the conformational changes during the isomerization result in the modulation of the amplitude and frequency of the high frequency fingerprint modes.

In this work, we apply (multidimensional) time resolved vibrational spectroscopic techniques like degenerate four wave mixing (DFWM),^{50, 52} pump-degenerate four wave mixing (pump-DFWM)^{53, 64-66} and pump-impulsive vibrational spectroscopy (pump-IVS)^{28, 58, 65, 67} to compare the vibrational dynamics of AT and 13C isomers of ASR. Tracking the excited state evolution of the nuclear wave packet shows (i) a delayed appearance of low frequency ($100\text{-}400\text{ cm}^{-1}$) modes and (ii) a blue shift for two characteristic high frequency modes: CH_3 rock ($\sim 1000\text{ cm}^{-1}$) and C=C stretch ($\sim 1500\text{ cm}^{-1}$) modes. The delayed rise of the low frequency ($100\text{-}400\text{ cm}^{-1}$) modes supports the indirect activation mechanism of the delocalized torsional modes by localized high frequency ($>1000\text{ cm}^{-1}$) modes, such as C=C stretch, via internal vibrational energy redistribution (IVR).^{50, 53} Experimentally observed blue frequency shifts during the course of isomerization hints at a reduction of the effective π -conjugation length due to the formation of non-planar 90° twisted state. Most remarkably, AT shows a slower frequency shift than 13C, which according to our interpretation, indicates AT reaches this twisted state much later than the 13C isomer. In addition, a relatively stronger amplitude of the HOOP mode in the GS of 13C compared to AT has been observed both in non-resonant DFWM, IVS and spontaneous Raman spectra which indicate that the GS of the 13C isomer is more pre-twisted than AT isomer of the RPSB in ASR.

Experimental methods and analysis

Sample preparation

ASR sample was prepared according to the standard reported protocol.⁴⁷ A purified sample was concentrated and dialyzed against a buffer solution containing 200 mM NaCl, 25mM Tris-HCl to maintain pH 7.0 and 0.01% DDM (n-Dodecyl- β -D-Maltopyranoside, Anagrade, Anatrace) was added to help the protein to get stabilized by forming micelles. For pump-probe experiments, the concentration of DDM was reduced (still above the CMC of DDM: 0.007%) compared to that used in the previous report,⁴⁷ in order to minimize the formation of bubbles while circulating it through the flow cell during the measurement.

Time-resolved experiments

The experimental setup used for the pump-DFWM and pump-IVS study is shown in Fig. S1 and S2, respectively. Briefly, a regeneratively amplified Titanium-Sapphire laser system (795 nm, 1 kHz) was used as fundamental laser source. The

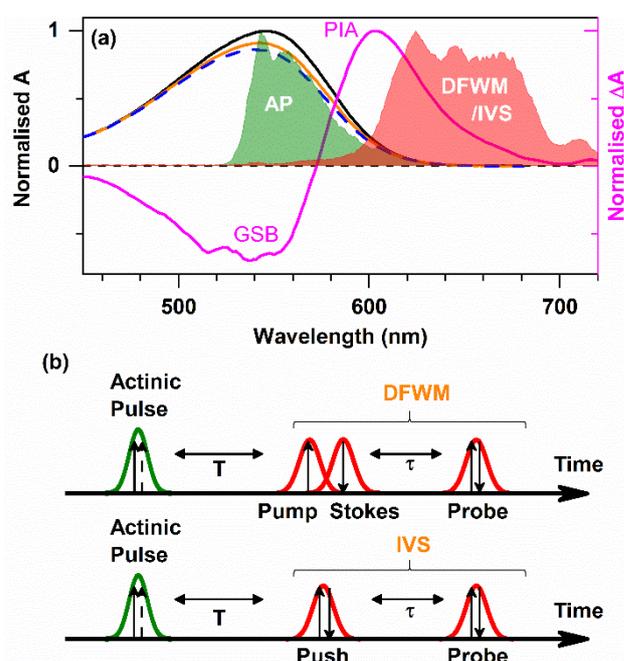


Figure 2: (a) Ground state absorption spectra of ASR in dark (AT) and light adapted (mixture of AT and 13C) conditions are shown by black and orange lines, respectively. The spectrum of pure 13C isomer (blue dotted line) was derived by taking linear combination of the dark and light adapted absorption spectra using known isomeric ratio. Normalized difference absorption (ΔA) spectra at pump-probe delay of 100 ps is shown by the magenta curve. Different characteristic bands: GSB (ground state bleach) and PIA (photo-induced absorption) are shown. Green and red (solid in (a) and line in (b)) curve represent the spectrum of the Actinic Pulse (AP) and degenerate pump/stokes/probe (DFWM) or push/probe (IVS) spectrum, respectively. (b) The pulse sequence used in pump-DFWM (top) and pump-IVS (bottom) experiments.

output was equally divided into two home built non-collinear optical parametric amplifiers (nc-OPA's) to generate the ultrashort actinic-pulse (AP) and degenerate pump/stokes/probe (DFWM) or push/probe (IVS) spectra (Fig. 2). In pump-DFWM and pump-IVS, the spectrum of the actinic pulse was spectrally resonant to the GS absorption (Fig. 2(a)), while the DFWM/IVS spectra overlapped with the photo-induced absorption (PIA) band of ASR (Fig. 2(b)). All pulses were compressed below 15 fs by a prism pair (Fig. S3).

The energy of the actinic pulse beam was attenuated to 100 nJ and focused to a spot with a diameter of 90 μm . The diameter of each of the DFWM/IVS beams was set to 50 μm and the energy to about 50 nJ. The delay between pump and push pulse (T) was controlled by a mechanical delay stage. For all measurements, the probe delay (τ) was scanned using a rapid scan approach. A single transient, with a length of 2 ps and time steps of 4 fs, was acquired in about 500 ms.

The optical density (OD) of the ASR sample was around 0.7 at 545 nm (λ_{max}). The sample was circulated through a fused silica flow cell with path length a 0.5 mm by a motor-driven pump to ensure that illuminated volume of the sample was replenished between two consecutive laser shots. The flow rate was optimized during the measurement in order to avoid the formation of bubbles. The sample was kept in dark overnight for dark adapted (DA) measurements. For light adaptation, the

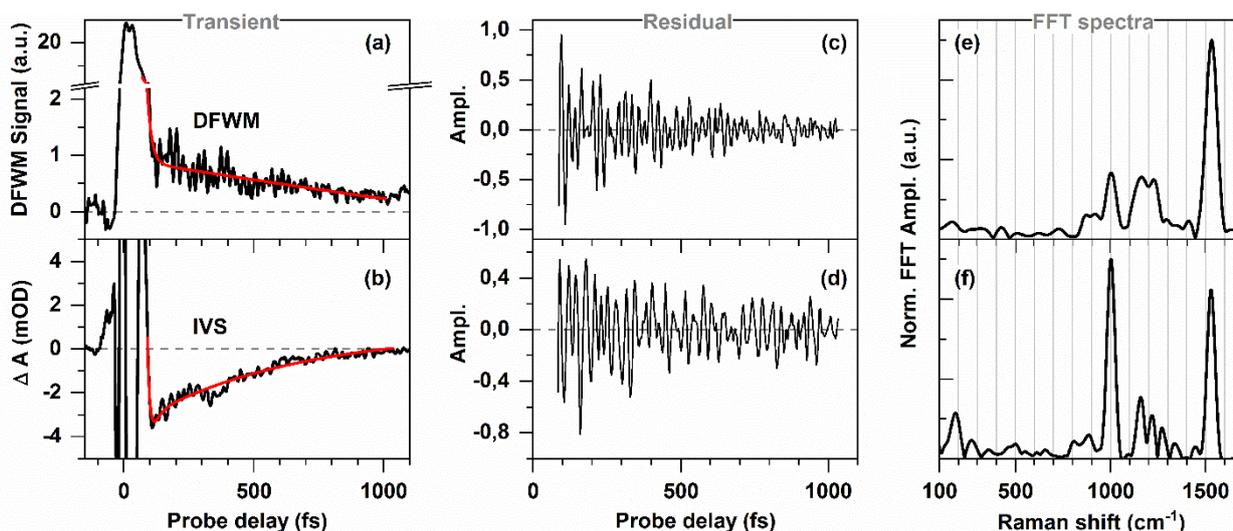


Figure 3: Transients obtained from (a) DFWM and (b) IVS measurements with the non-resonant DFWM/IVS excitation spectrum (i.e. in absence of actinic pulse) at detection wavelength 590 nm under dark adapted condition. The non-oscillatory signal was subtracted in each case by a bi-exponential fitting (red line) before FFT. The corresponding oscillatory signal, obtained after fitting DFWM and IVS transients, are shown in (c) and (d) respectively. Here, a Butterworth filter has been applied on the residuals to cut-off the low frequency component below 100 cm^{-1} in order to suppress noise. The corresponding FFT spectra obtained in DFWM and IVS experiments are shown in (e) and (f) respectively.

sample was illuminated for 1 hour with 500 mW from a LED source (Luxeon LXHL-PL01) centered at 590 nm. During the measurement of light adapted (LA) samples, the transparent sample reservoir was exposed to the same LED source to maintain LA photo-stationary state and it was covered with black paper during the measurement of DA. The GS absorption spectrum was measured before and after each set of time resolved measurements to ensure the isomeric ratio remains the same and also to check if any degradation of the sample happened during the span of a measurement (about 20 min).

Signal analysis

The non-oscillatory contributions in each of the DFWM and IVS transients were subtracted by a bi-exponential fitting (Fig. 3(a)-(b)). In each case, 1 ps of the transients was fitted, leaving out the initial 80 fs to avoid the coherent artefact. A typical algorithm⁶⁶ was followed to convert the remaining oscillatory signals (Fig. 3(c)-(d)) from time domain to the spectral domain (Fig. 3(e)-(f)). The residuals were multiplied by a Gaussian window and subsequently zero padding was done before performing a fast Fourier transformation (FFT).

The signal to noise ratio in (pump-) DFWM measurements was almost one order of magnitude better than that in (pump-) IVS measurements. For (pump-) DFWM measurements, each of transient was averaged 60 times whereas it was averaged 600 times for (pump-) IVS measurement to achieve a comparable S/N ratio. This difference in the S/N ratio is because the self-heterodyne IVS signal suffers from the fluctuation of intensity of the local oscillator probe whereas the DFWM signal, being a homodyne technique, is background free.

Figure 3 also shows the well-known effect of the homodyne detection in (pump-)DFWM on the decay of the oscillatory signal:⁶⁵ Oscillatory contributions in the residual decay faster in the DFWM signal compared to IVS signal. The intrinsic interference between population grating and vibrational

coherence results in a faster decay of the oscillation and consequently makes the FFT spectra broader (compare Fig. 3(e) and (f)). This also causes the small deviations in the spectral peak positions between DFWM and IVS FFT spectra (see e.g. Fig. 4), especially for those vibrational modes (e.g. $1100\text{--}1400\text{ cm}^{-1}$) which are very closely spaced.

Experimental results

Non-resonant DFWM/IVS experiments

DFWM and IVS experiments with almost non-resonant DFWM/IVS spectra (Fig. 2(a)) were carried out to capture the pure GS vibrational spectra. The FFT spectra obtained after the subtraction of non-oscillatory contributions from the transients, show the activity majorly in the high frequency region ($>1000\text{ cm}^{-1}$) for both DFWM (Fig. 4(a)-(d)) and IVS (Fig. 4(e)-(h)) experiments. A weak activity in the low frequency region ($<400\text{ cm}^{-1}$) was also observed, especially at the detection wavelength at 630 nm, where the relative amplitudes of the high frequency ($>1000\text{ cm}^{-1}$) modes are lowered. This variation of relative intensity of the low and high frequency modes from the edge (590 nm) to the center (630 nm) detection wavelength of DFWM/IVS spectrum are in agreement with the natural spectral dependence of CARS-based schemes.⁶⁸

The comparison between the FFT spectra of DA and LA ASR shows mainly three characteristic features. Firstly, the modes which appear at 1003 and 1530 cm^{-1} in the DA DFWM FFT spectrum (Fig. 4(a) and (c)) shift to 1007 and 1539 cm^{-1} (Fig. 4(b) and (d)), respectively, for LA ASR. Secondly, the DFWM FFT spectrum of DA ASR shows two peaks around 1165 and 1230 cm^{-1} , while for LA ASR they appear at 1180 and 1300 cm^{-1} . All these changes are also observed in IVS measurements (compare Fig. 4(e) and (g) to (f) and (h)). The modes around $1100\text{--}1400\text{ cm}^{-1}$ also show in IVS qualitatively similar contrasts between DA and LA ASR as observed in DFWM. Three peaks at

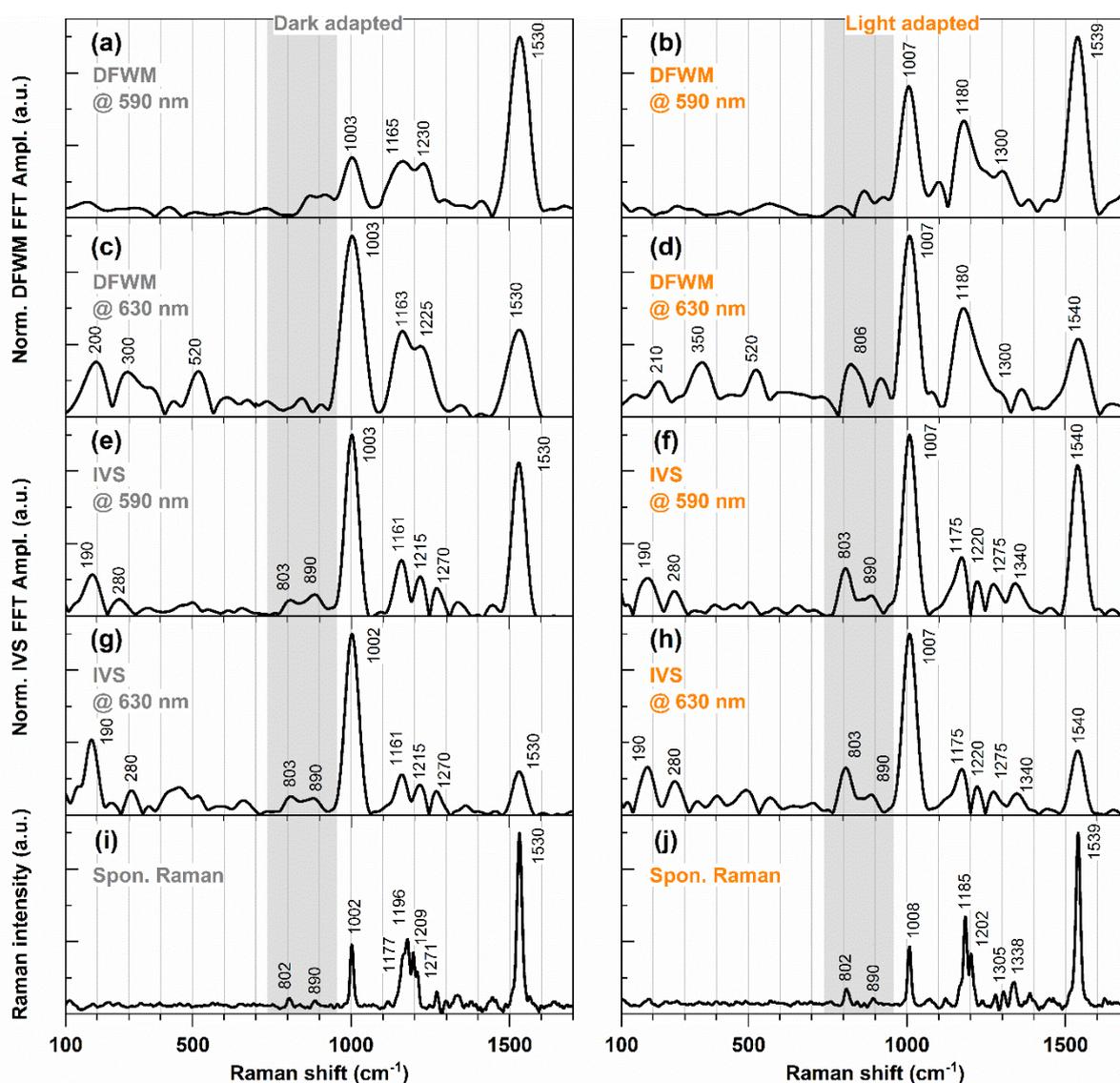


Figure 4: The FFT spectra obtained after subtraction of the non-oscillatory components from the transients of DFWM (a-d) and IVS (e-h) experiment probed at 590 nm (a, b, e, f) and 630 nm (c, d, g, h). The left and right column represent the FFT spectra of DA and LA ASR respectively. The spontaneous Raman spectra of DA and LA ASR have been shown on the bottom graphs i and j respectively. The grey shaded area (750-950 cm^{-1}) represents the characteristic frequency region of the HOOP modes of RPSB.

1161, 1215 and 1270 cm^{-1} for DA ASR (Fig. 4(e) and (g)) change to 1175, 1220, 1270 and 1340 cm^{-1} under LA condition (Fig. 4(f) and (h)). Finally, the modes appearing in the region of 750-950 cm^{-1} (grey shaded area in Fig. 4), in particular the mode at about 806 (DFWM)/ 803 (IVS) cm^{-1} , shows relatively stronger amplitude for LA ASR than for DA ASR. This is a very important result which will be discussed later.

The observed GS fingerprint vibrational modes of RPSB have been already previously assigned.^{7,44,69-74} We hereby follow the same assignment of the main spectral peaks to specific vibrational motions (Table 1). For example, the mode around 1000-1010 cm^{-1} is assigned to CH_3 rock and that around 1530-1540 cm^{-1} is assigned to C=C stretching mode. The multiple peaks in the region from 1100 to 1400 cm^{-1} are mostly known as the signature of the stretching vibration of different C-C bonds present in RPSB, sometimes coupled with C-C-H

in-plane-rock modes. In addition, the low frequency modes (190, 300 cm^{-1}) are usually assigned to the delocalized torsional modes whereas the Raman activity in the region 750-950 cm^{-1} (grey shaded area in Fig. 4) is well known for the HOOP wags (Table 1). In addition, a non-resonant ($\lambda_{\text{exc}}=785$ nm) spontaneous Raman measurement (Fig. 4(i)-(j)) was performed to further corroborate the vibrational spectra obtained from the time resolved experiments (DFWM and IVS). While the non-resonant Raman results match very well most of the frequencies and respective shifts observed with DFWM/IVS (Fig. 4(a)-(h)), there are three major contrasts: (i) Different frequencies for modes in the spectral region between 1100 and 1400 cm^{-1} , (ii) Complete absence of low frequency modes (<750 cm^{-1}) in the non-resonant Raman measurements (which were active in the DFWM/IVS measurements) and (iii) Different amplitude of the 803/805 cm^{-1} mode.

Table 1: Comparison of the central frequencies (in cm^{-1}) of the spectral peaks obtained from non-resonant DFWM, IVS and spontaneous Raman measurements. Assignment of the modes were done based on the literature.

Dark adapted (98% AT)			Light adapted (64% 13C; 36% AT)			Assignment of the modes
DFWM	IVS	Spon. Raman	DFWM	IVS	Spon. Raman	
200, 300, 520	190, 280	-	210, 350, 520	190, 280	-	Delocalized torsion ^{71,74}
900	803, 890	802, 890	810, 900	803, 890	802, 890	Hydrogen out of plane wag ^{7,44,73}
1003	1003	1002	1007	1007	1008	CH_3 rock ^{7,69,70,73}
1165, 1230	1161, 1215, 1270	1177, 1196, 1209, 1271	1180, 1300	1185, 1202, 1305, 1338	1175, 1220, 1275, 1340	C-C stretch + C-C-H in plane rock ^{7,69,70,73}
1530	1530	1530	1530	1539	1539	C=C stretch ^{7,69,70,73}

Separation of Isomer Specific Contribution

While the spectral features observed for DA and LA ASR (Fig. 4) already mirror the different isomer concentrations, in the following a quantitative extraction of the pure AT and 13C spectra will be performed. In our previous report,⁷⁵ we successfully showed the separation of the GS isomer specific contributions for two individual high frequency modes (CH_3 rock and C=C stretch) by a single Gaussian fit. A Gaussian model was used instead of a Lorentzian model, since the former showed a better fitting quality. Here, we expand and present a more global approach (Fig. 5) to fit all the modes appear in the spectral region from 950-1600 cm^{-1} by a series of Gaussians:

$$FFT_{AT\ GS} = \sum_i^n a_i * G_i(v_i, \omega_i) \quad (1)$$

where each Gaussian (G_i) represents an individual vibrational mode. The amplitude (a_i), central frequency (v_i) and width (ω_i) of each Gaussian was varied in this multi-Gaussian fit (equation 1). Since DA ASR contains exclusively the AT isomer

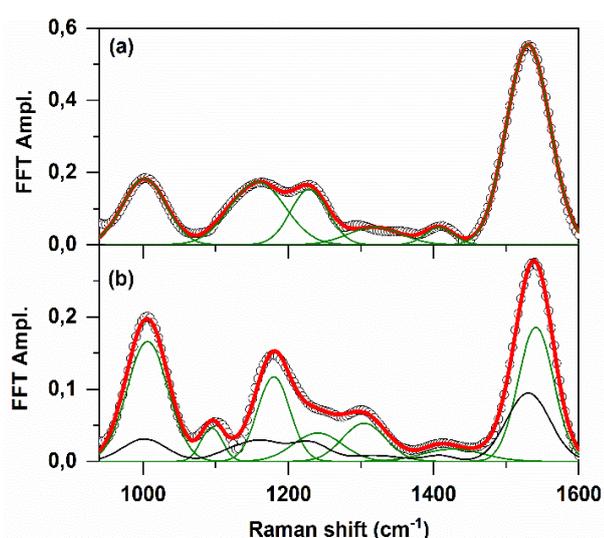


Figure 5: Multi-Gaussian fit of (a) DA and (b) LA non-resonant DFWM FFT spectra probed at 590 nm. The series of Gaussians represented by green curves in each graph. Black line in (b) represent the spectra of AT isomer which is kept constant during this constrained fit.

(98%),^{46,47} the multi-Gaussian fit (Fig. 5(a)) gives the parameters (a_i, v_i, ω_i) for each of the vibrational modes of AT ASR. This information ($FFT_{AT\ GS}$) can be used to disentangle the pure spectrum of 13C GS by a constrained fit:

$$FFT_{LA\ GS} = \sum_j^n a_j * G_j(v_j, \omega_j) + f * FFT_{AT\ GS} \quad (2)$$

Here, the fitted spectra ($FFT_{AT\ GS}$) obtained from the previous multi-Gaussian fit was kept constant and the parameters (a_j, v_j, ω_j) of a new set of Gaussians, which represent the modes of the 13C isomer, were varied to fit the LA non-resonant DFWM FFT spectrum (Fig. 5(b)). The factor f in equation (2) depends on the percentage (36%)⁴⁶ of the AT isomer presented in the LA form. Since the DFWM signal is directly proportional to the square of the sample concentration, f should be equal to the square of the fraction of AT isomer present in LA ASR: $(0.36)^2=0.13$. The fitted value obtained for the factor ($f=0.14 \pm 0.02$) matches this value

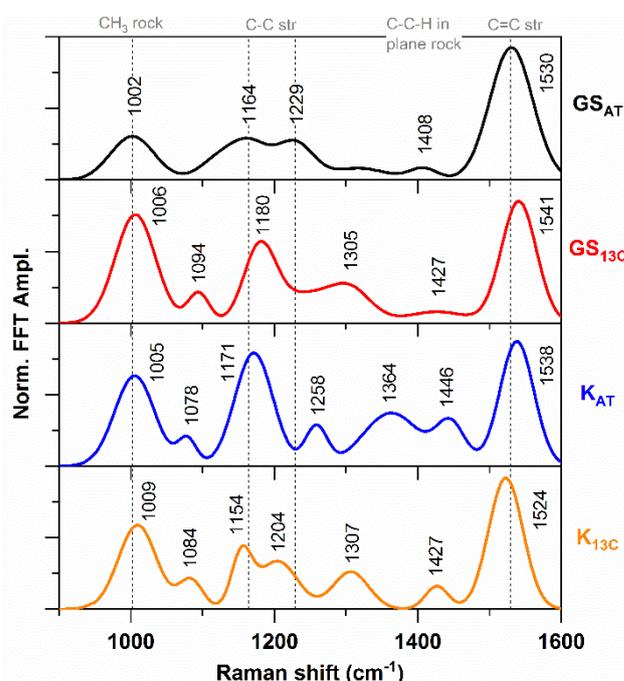


Figure 6: The separated pure spectra of AT GS, 13C GS, K_{AT} and K_{13} are represented by black, red, blue and orange line, respectively. The black dotted lines represent the central frequencies obtained for AT GS.

very well. The pure spectrum of the ground state of the 13C isomer is shown in Fig. 6.

The same approach can also be applied to obtain the spectrum of the K-photoproduct of each isomer (K_{13C} and K_{AT} , see Fig. 1). In order to obtain the DFWM signal of these photoproducts, each isomer was directly excited by an actinic pulse spectrally resonant with the respective GS absorption. The DFWM spectrum was still spectrally resonant with the photo-induced absorption (PIA) band (see Fig. 2) but delayed at $T=100$ ps after the AP. This particular delay was chosen as the formation of the vibrationally thermalized K-photoproduct (Fig. 1) is known to take place within 100 ps.^{46, 47} The pump-DFWM signal in this condition contains several contributions (Fig. 1). The signal of DA ASR after 100 ps, for example, contains contributions of the AT GS and of K_{AT} , which are reasonably easy to separate in pump-DFWM and less in pump-IVS due to noise (not shown). The pump-DFWM signal of the LA ASR, however, is much more challenging: At 100 ps delay, it contains the signal of the GS of both isomers, as well as from both photoproducts. In order to extract the K_{13C} , the spectra of AT, K_{AT} and 13C must be used, which is inherently more susceptible to noise. With this information in mind, the pure spectra of four different stereo-isomers (GS_{AT} , GS_{13C} , K_{AT} and K_{13C}) appearing in the photo-cycle are depicted in Fig. 6 (see also

Fig. S5). Several differences in spectral signatures of these four species can be observed. For example, the C=C stretching mode of 13C GS (1541 cm^{-1}) gets blue shifted compared to that of AT GS (1530 cm^{-1}). Also, isomerization at $C_{13}=C_{14}$ position causes the C=C stretching mode to shift in the higher frequency from AT GS (1530 cm^{-1}) to K_{AT} (1538 cm^{-1}), whereas it shifts to the lower frequency from 13C GS (1541 cm^{-1}) to its corresponding sub-ns photo-product K_{13C} (1524 cm^{-1}). Similar trend is also observed for the CH_3 rock mode while for K_{13C} , it gets blue shifted compared to 13C GS. The frequencies and amplitudes of modes in the spectral region between 1100 and 1300 cm^{-1} are very different for each isomer. Finally, it is important to note that the spectral peaks appearing in the region 1300 – 1450 cm^{-1} , which have been assigned to C-C-H in plane rock mode (Table 1), do not change from 13C GS ($1305, 1427\text{ cm}^{-1}$) to K_{13C} ($1307, 1427\text{ cm}^{-1}$), whereas a significant change is observed from AT GS to K_{AT} . A single weak peak at 1408 cm^{-1} appears for AT GS whereas at least two strong peaks appear for K_{AT} ($1364, 1446\text{ cm}^{-1}$) in this region.

Pump-DFWM experiments in the sub-ps timescale

In order to follow the isomerization reaction of both 13C and AT isomers in the excited state, pump-DFWM measurements were

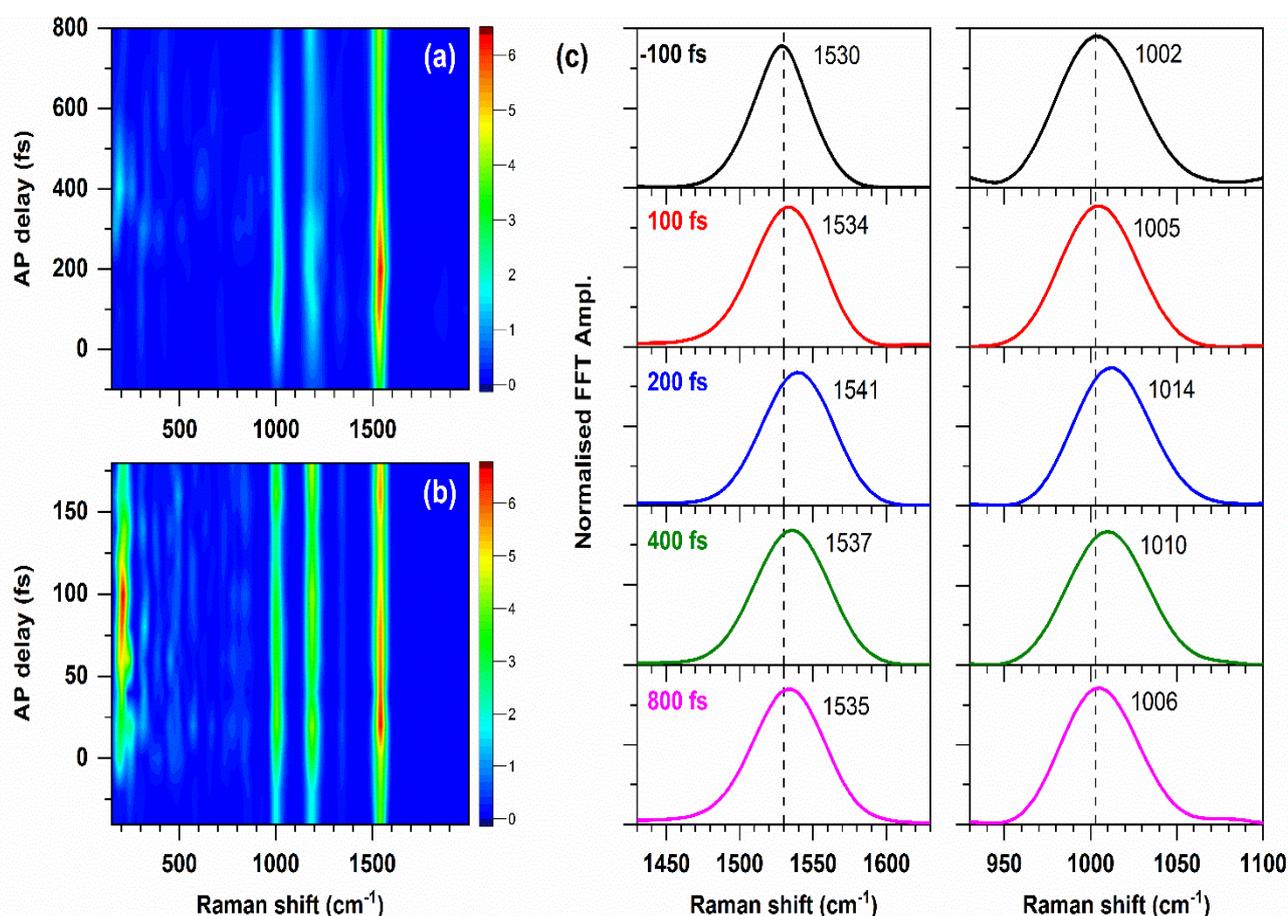


Figure 7: The evolution of pump-DFWM FFT spectra probed at 590 nm for (a) DA and (b) LA ASR with the actinic pulse delay. (c) The frequency shift of C=C stretch (left) and CH_3 rocking (right) mode for DA ASR with different actinic pulse delay. The black dotted lines represent the central frequencies obtained for AT GS (Fig. 6). The frequency shifts of the corresponding modes in LA ASR are shown in Fig. S6 of ESI.

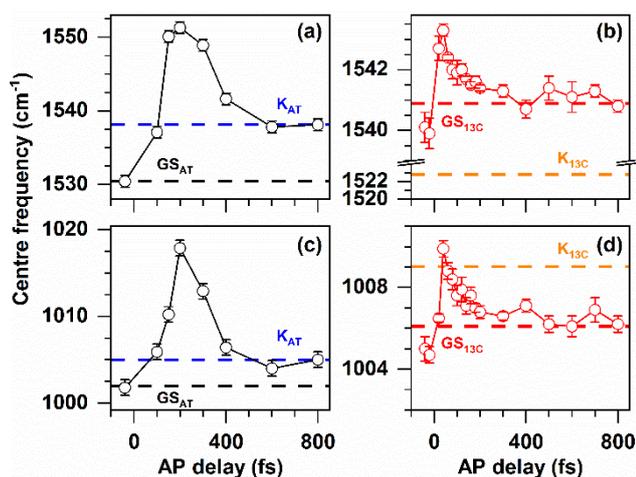


Figure 8: The excited state evolution of C=C stretching (a, b) and CH₃ rock modes (c, d) obtained for the pure AT isomer (a, c) and LA (b, d) form ASR. The central frequencies of pure GS (AT and 13C) and K-photo intermediates (K_{AT} and K_{13C}) are shown by black, red, blue and orange dotted lines respectively in each graph. Probe detection wavelength was 590 nm.

performed by varying the actinic pulse delay (T) up to about 1 ps (Fig. 7). Fourier-transformed spectra for DA and LA are displayed in Fig. 7(a) and (b), respectively. In both situations, two common changes can be observed after the arrival of the actinic pulse ($T > 0$): (i) amplification of the high frequency modes (> 1000 cm⁻¹) and (ii) appearance of the strong low frequency modes (100–400 cm⁻¹), in particular for LA ASR (Fig. 7(a)), which were very weak in the GS (Fig. 4). Moreover, a closer look on the high frequency (> 1000 cm⁻¹) modes (Fig. 7(c) and Fig. S6) shows a significant spectral shift with the actinic pulse delay. The major spectral shifts of DA and LA ASR happen within 800 and 200 fs, respectively, which match respective excited state lifetimes^{46, 47} of the AT (750 fs) and 13C (120 fs) isomers. The frequency evolution for each of the characteristic vibrational modes are described in detail in the following.

C=C stretching and CH₃ rock mode: Figure 7 (c) depicts the evolution of C=C stretch and CH₃ rock modes for DA ASR. Here, a negative time delay means the actinic pulse comes after DFWM interaction which should basically give the GS vibrational spectra. Hence, the central frequencies of C=C stretch and CH₃ rock modes at $T = -100$ fs, which appear at 1530 and 1002 cm⁻¹, match the GS frequencies displayed above (Fig. 4). At initial positive T -delays (until 200 fs), each of these modes shows a rapid blue shift and subsequently a slower red shift (see DA in Fig. 7(c) and Fig. S7(a) and (c)). Although the FFT spectra at $T > 0$ contains the major contribution of the transient species being resonantly probed at the corresponding time, a minor contribution of GS coherence, due to the non-resonant Raman interaction, still contaminates the signal. As a consequence, pump-DFWM FFT spectra at any positive actinic pulse delay (T) is not the pure spectra of the transient species formed within the corresponding time (T) after the initiation of the reaction by AP. Since the pure GS spectra of AT isomer is already known (Fig. 5 and 6), it is possible to extract the pure spectra of the transient species present at different actinic pulse delays by the

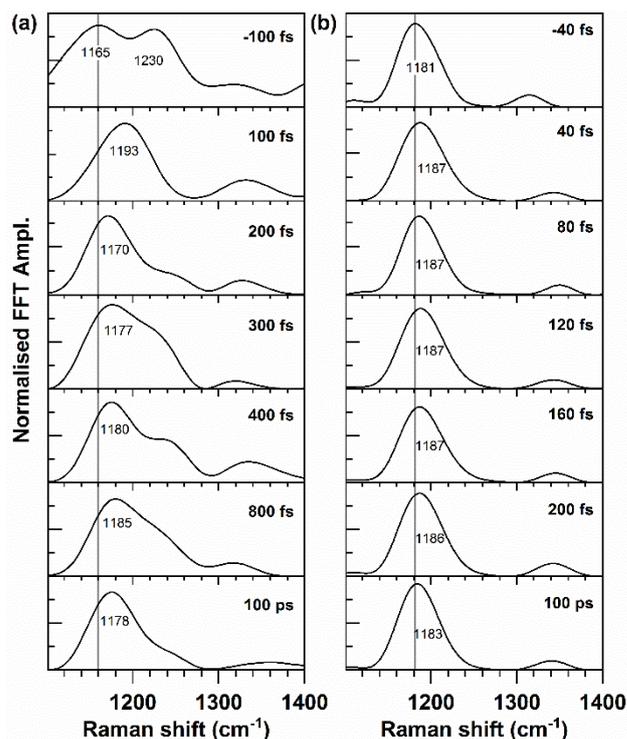


Figure 9: The evolution of fingerprint (1100–1400 cm⁻¹) modes in the pump-DFWM FFT spectra for (a) AT and (b) LA ASR at different actinic pulse delay (T) probed at 590 nm. The vertical lines in (a) and (b) represent the corresponding central frequencies of AT and 13C GS, respectively.

aforementioned constrained multi-Gaussian fitting (Fig. 5 and Fig. S5).

By applying this method, the frequency of C=C stretch and CH₃ rock modes for AT is shown in Fig. 8 (a) and (c), respectively. AT shows a clear strong blue shift of 21 and 18 cm⁻¹ for the C=C stretch and CH₃ rock modes, respectively, at $T = 200$ fs compared to that of GS species ($T < 0$). At longer delays ($T > 200$ fs), it undergoes a red shift to 1538 and 1005 cm⁻¹. These two latter frequencies are the central frequencies of the C=C stretch and CH₃ rocking modes, respectively, of previously extracted pure K_{AT} (Fig. 6). The separation of all contributions to the transient LA ASR signal is more challenging than for DA ASR. Here, two ground state species are excited (AT and 13C), followed by two excited states and two photoproducts, which leads to much larger frequency uncertainties and ambiguous results. Therefore, the frequency shifts of the C=C stretch and CH₃ rock modes for the LA ASR are shown without extraction (Fig. 8(b) and (d), respectively). A similar trend is observed as for AT, i.e. an initial blue shift takes place followed by a red shift. However, the blue shift is much smaller and the whole dynamics is much faster in the LA than for AT (compare e.g. Fig. 8(a) to (b)). The maximum of the blue shift appears at much earlier actinic pulse delay (about $T = 40$ fs) compared to AT form (about $T = 200$ fs).

Fingerprint (1100–1400 cm⁻¹) region: As observed for the GS spectra (Fig. 4, 6), the fingerprint region around 1100–1400 cm⁻¹ (Fig. 9) shows a congested spectrum after the actinic excitation, particularly for AT ASR. The double peak feature (1165 and 1230 cm⁻¹) before time zero merges to single major peak

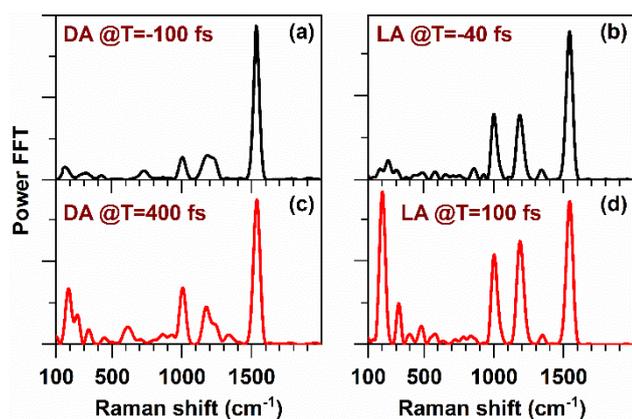


Figure 10: The relative amplification of low frequency modes probed at 590 nm before (black, $T < 0$) and after (red, $T > 0$) the arrival of actinic pulse for DA (a, c) and LA (b, d) ASR.

(1193 cm^{-1}) along with a minor peak (1330 cm^{-1}) at $T=100$ fs for AT ASR. Afterwards, the major peak shows a red shift till 200 fs which is followed by, a slow monotonous blue shift until the formation (200 to 800 fs) of the hot-intermediate, commonly known as J-species in literature. Subsequently, it undergoes a small red shift during vibrational relaxation on a much longer time scale (1 to 100 ps) to form the thermally relaxed photoproduct (K_{AT}). This contrasts to the signal of LA ASR, where the major peak at ~ 1181 cm^{-1} undergoes a blue shift initially ($T=40$ fs) to 1187 cm^{-1} and does not shift afterwards within the excited state lifetime (< 120 fs) of the 13C isomer. It only shows a small (1187 to 1183 cm^{-1}) red shift during the photoproduct vibrational relaxation (~ 1 ps to 100 ps), similar to the AT isomer.

Low frequency (< 400 cm^{-1}) torsion and HOOP modes: As shown in Fig. 4, the low frequency modes below 400 cm^{-1} are absent or very weak in GS. However, a delayed activation of these low frequency modes (Fig. 7(a) and (b)) is observed after the excitation with the AP. This is in contrast to the activity of the high frequency modes, whose amplitude rises as soon as the AP arrives (Fig. S8). This contrasting activity of low frequency (< 400 cm^{-1}) modes is further evidenced in Fig. 10 at several actinic pulse T-delays. A significant relative amplification of the low frequency modes (190, 300 cm^{-1}), compared to the high frequency (> 1000 cm^{-1}) modes, is observed at $T > 0$ for DA and LA ASR. Here it is important to note that LA ASR shows a much stronger low frequency activity than DA ASR.

The evolution of the amplitude of the low frequency modes is different from other modes. While high frequency modes (e.g. C=C stretch) show an instantaneous rise in the activity at very early actinic pulse T-delays, low frequency (e.g. 190 cm^{-1}) activity shows a delayed rise for LA ASR (Fig. 7). The FFT amplitude of the low frequency (190 cm^{-1}) modes takes about 100 fs to reach the maxima whereas that of C=C stretching mode reaches the maxima within the experimental time resolution after the actinic excitation ($T=20$ fs). In addition, the exponential rise time (70 ± 20 fs) of low frequency (190 cm^{-1}) modes matches the exponential decay time (60 ± 40 fs) of C=C stretch mode (Fig. S8). Moreover, no significant evolution of

HOOP modes (800-1000 cm^{-1}) is observed in our measurements. This will be further discussed and investigated later.

Discussion

Ground state Raman activity

A strong Raman activity of the vibrational modes mainly in the high frequency (> 1000 cm^{-1}) region of the spectra has been observed in non-resonant experiments (Fig. 4). The convolution of the GS absorption spectrum with the non-resonant DFWM/IVS excitation spectrum (Fig. 2), gives a FWHM of about 500 cm^{-1} and, therefore, there is a negligible induction probability of a vibrational coherence above 500 cm^{-1} in the ES potential surface. Hence, all modes with frequencies well over 500 cm^{-1} in these non-resonant measurements, including HOOP, CH_3 -rock, C-C and C=C modes, are assigned to GS manifold. This is further supported by the spontaneous Raman spectra (Fig. 4(i) and (j)), which show a good correlation with the FFT spectra obtained from time resolved experiments for frequencies well over 500 cm^{-1} . However, the weak but detectable Raman activity observed in the low frequency region (200 and 300 cm^{-1}) in both DFWM and IVS signal, does not appear in any spontaneous Raman spectra. If this low frequency activity originates from GS, it should definitely be visible in the spontaneous Raman measurements. Therefore, we assign all low frequency (< 400 cm^{-1}) activity detected with DFWM/IVS to the excited state manifold. A detailed discussion about the origin and activation mechanism of these mode will be presented in next section.

The extraction of the pure GS spectra of AT and 13C isomer (Fig. 6) enables us to make a quantitative comparison. The central frequencies of C=C stretch and CH_3 rock modes are shifted about 11 cm^{-1} (1530 vs 1541 cm^{-1}) and 4 cm^{-1} (1002 vs 1006 cm^{-1}), respectively, between AT to 13C ASR. This is in good agreement with reported values for ASR.⁴⁶ A similar increase of high frequency modes from all-*trans* to *cis* isomer has also been reported recently for chanello-rhodopsin⁵⁸ and visual rhodopsin,⁴⁹ which further corroborates the separation of the

Table 2: Comparison of the selected fingerprint vibrational modes for all-*trans* and *cis* isomers found for ASR in this report and those reported for retinal protonated Schiff base (RPSB) in solution and different proteins: bacteriorhodopsin, visual rhodopsin and chanello-rhodopsin.

Sample	CH_3 rock [cm^{-1}]	C-C stretch + C-C-H in plane [cm^{-1}]	C=C stretch [cm^{-1}]
AT-RPSB in methanol ⁵²	1010	1160, 1205	1565
Bacteriorhodopsin (AT) ⁵⁰	1008	1165, 1210	1530
Visual rhodopsin (AT) ⁴⁹	-	1167, 1322	1541
Visual rhodopsin (11- <i>cis</i>) ⁴⁹	-	1173, 1275, 1313, 1363	1550
Chanello-rhodopsin (AT) ⁵⁸	1011	1161, 1208, 1281	1531
Chanello-rhodopsin (13- <i>cis</i>) ⁵⁸	1017	1157, 1196, 1301, 1369	1545
ASR (AT) ^{This report}	1002	1164, 1229	1530
ASR (13- <i>cis</i>) ^{This report}	1006	1094, 1180, 1305	1541

pure spectra of ASR. Furthermore, the C-C stretch region shows multiple distinguishable peaks between AT and 13C ASR (Table 1). Two peaks around 1165 and 1230 cm^{-1} , observed for AT ASR (Fig. 6), are quite common for the retinal chromophores in AT conformation in other proteins (Table 2). On the other hand, the appearance of a distinguishable mode above 1300 cm^{-1} for 13C ASR is a general marker of *cis* form (Table 2).⁴⁹ In addition, the C-C stretch mode which appears at 1180 cm^{-1} in case of 13C ASR, has been attributed as an indicator for the formation of 13C-isomer of BR.⁷⁶

The activity of HOOP modes (800-1000 cm^{-1}) in GS (grey shaded region in Fig. 4) has been interpreted as an indirect indicator of a distorted non-planar structure of RPSB.^{71, 77} The relative amplitude of the HOOP modes, particularly at about 805 cm^{-1} , is higher in LA (major component is 13C) ASR compared to DA (98% AT) ASR. This observation is consistent in each (DA vs LA) of the spectra measured by three different spectroscopic techniques (DFWM, IVS and spontaneous Raman). All these together, hints at the presence of a non-planar structure for the 13C isomer, although it is much less distorted than the 11-*cis* isomer in visual rhodopsin.^{44, 69, 71, 77} A detailed assignment of H-wag modes is, however, necessary in order to specify the region of distortion in the long retinal chain. The complete assignment of H-wag modes for ASR has not been reported yet but it is well known for BR from the resonant Raman study by Smith *et al.*⁷⁰ In that report, a mode at 800 cm^{-1} was assigned to C₁₄-H out-of-plane wag and showed strong amplitude for 13C isomer whereas it almost disappears for AT isomer, similar to ASR isomers in our study. Later, solid state NMR data^{78, 79} indicated a difference in the structural rigidity along the C₁₃=C₁₄-C₁₅ moiety between AT and 13C isomers inside the retinal pocket of BR. Following the same line, our observation can be interpreted as both isomers are pre-twisted around or close to C₁₃=C₁₄ bond of retinal chromophore but 13C isomer is significantly more distorted than AT.

The exact position of this distortion and the differences between the two isomers can be further numerically investigated by applying a classical (harmonic) atomistic force field and performing molecular dynamics simulations of ASR embedded in a membrane model (see ESI). The geometry of AT and 13C GS inside the retinal pocket shows that both isomers are equally distorted around the C₁₃=C₁₄ bond, i.e. the dihedral angle of C₁₂-C₁₃-C₁₄-C₁₅ is found to be +191.6° and 11.7° (Table S1) for AT and 13C isomer, respectively. This is in good agreement with the value (~13°) previously reported by QM/MM geometry optimization calculations.³¹ These new calculations, however, shows that the isomers significantly differ around the neighboring C₁₄-C₁₅ bond (dihedral C₁₃-C₁₄-C₁₅-N): While AT is nearly planar (177.1°) around the C₁₄-C₁₅ bond, the 13C is almost 10° twisted (189.5°). This relatively larger pre-twist around C₁₄-C₁₅ bond for 13C isomer corroborates very well the experimental observation of higher amplitude of C₁₄-H wag as discussed above in the previous paragraph.

Finally, a brief comparison of vibrational signatures of GS isomer with its corresponding K-photoproducts (Fig. 6) helps to further rationalize the frequency shifts in terms of conformational differences. The conformational changes alter

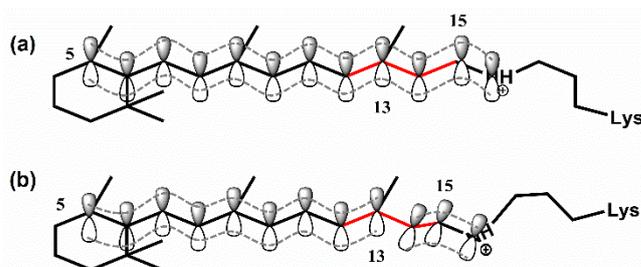


Figure 11: The reduction of the effective π -conjugation length from (a) Franck-Condon excited state to (b) twisted transition state (far Franck-Condon region). In (a), there is an extended π -conjugation from C5-atom to N-atom of the Schiff base which is reduced in (b).

the delocalization of δ and π -electron density for the macro-molecule like RPSB, which explains the frequency shift of CH₃ rock, C-C and C=C stretch and C-C-H in-plane rock modes. The changes in the frequency and amplitude of C-C-H in-plane rock (1300-1400 cm^{-1}) modes (Fig. 6) are important as these are reported to be affected by the torsion around the active C=C bond.⁶² Although the frequency of this mode changes from AT GS (1408 cm^{-1}) to K_{AT} (1364, 1446 cm^{-1}), it remains almost the same for 13C GS (1305, 1427 cm^{-1}) and for K_{13C} (1307, 1427 cm^{-1}), indicating a very similar strain around the C₁₃=C₁₄ bond for the latter two isomers. These results also corroborate FTIR results which illustrated that isomerization causes larger rotation around the active C₁₃=C₁₄ bond in the AT compared to the 13C isomer.⁷³

Excited state evolution of high frequency modes ($\geq 1000 \text{ cm}^{-1}$)

The pump-DFWM measurements captured the sub-ps frequency shifts of high frequency modes. In general, the C=C stretch and CH₃ rock modes showed a blue shift and subsequent red shift for both AT and LA (major 13C) ASR. Since the DFWM spectra, used for this measurement, covers the region (580-720 nm) where both excited state absorption and stimulated emission of ASR overlaps,^{46, 47} there are two possible origins for the observed frequency shifts: (i) the excited state evolution of the nuclear wave packet or (ii) the vibrational relaxation of the GS wave packet, generated by stimulated emission pumping (SEP) process.^{24, 38, 66} For the latter, we would expect a red shift just after the excitation for an anharmonic ground state potential.²⁴ This is because the GS wave packet, generated via SEP, in the higher lying vibrational states is lower in frequency than a relaxed GS wave packet, generated via non-resonant ISRS in the lower lying vibrational states. On the contrary, we observed a blue shift compared to the GS. This excludes this possibility and hence, the frequency shift is interpreted as coming mostly due to the excited state wave packet motion. In addition, the time scale of the frequency shifts is significantly different for AT and 13C isomers. It matches very well the excited state lifetime of each isomer, which further indicates that the observed frequency shifts originate due to the transient evolution of the ES species.

Sub-ps transient frequency shifts in the fingerprint region have been frequently interpreted as the conformational

changes associated with the isomerization process.^{62, 63} Here, the blue shift of C=C stretch (Fig. 8) is interpreted as the reduction in conjugation length due to the rotation around the C₁₃=C₁₄ bond during the isomerization. After the excitation by the actinic pulse, the RPSB is promoted to the Franck-Condon point of the excited state where it is still in the same geometry as it is in ground state (Fig. 11(a)). In this geometry, an extended π -conjugation is present along the retinal chain by the p_z-orbitals from C₅ up to protonated Schiff base N atom. As the isomerization reaction progresses, the RPSB starts to adopt a more non-planar twisted structure (Fig. 11(b)) towards the photoproduct geometry. During this evolution, the C₁₄-C₁₅-N π -conjugated moiety becomes almost perpendicular to the remaining π -conjugated system (C₅ to C₁₃) at the 90° twisted form (AT*) and the conjugation between two moieties, mentioned before, gets completely disrupted. Hence, the effective conjugation length gets reduced during evolution from the FC towards the 90° twisted form. It is well known for linear conjugated polyenes that the reduction of effective conjugation length causes a frequency blue shift of the C=C stretch mode.^{80, 81} Thus, decrease in the effective conjugation length explains the blue shift of C=C stretching mode observed here for retinal in ASR. The blue shift of CH₃ rock mode can also be interpreted in a similar way. The disruption of the C₁₃=C₁₄ bond during isomerization reduces the partial positive charge over the C₁₃ atom induced by the positively charged Schiff base N-atom. The partial positive charge over C₁₃ tends to pull the electron density from the CH₃ moiety (bonded to C₁₃) towards it but the lack of inductive effect in the twisted state pushes the electron density towards the CH₃ moiety. This possibly causes the blue shift of the rock mode of CH₃. This also corroborates the picture of change in electron density during the isomerization of BR, recently captured by the transient femtosecond X-ray spectroscopy.²⁹ Thus, following our argument, the slower frequency shift observed for AT isomer (Fig. 8(a)) compared to 13C isomer (~LA, Fig. 8(b)) can be interpreted as it takes longer

for the AT isomer to rotate around the C₁₃=C₁₄ bond to form the 90° twisted state compared to 13C. This corroborates as well QM/MM calculations³¹ which predict a barrier in the excited trajectory of AT and, therefore, a slower formation of the respective twisted transition state. Furthermore, the subsequent slow red shift for both isomers is a signature of increase in π -conjugation as the RPSB structurally changes from the 90° twisted species to the *relatively* more planar initial GS or photoproduct.

The frequency shift of the fingerprint modes in the region from 1100-1400 cm⁻¹ are relatively more complicated than the one observed for the C=C stretching mode. This is because different C-C stretching modes (C₈-C₉, C₁₀-C₁₁, C₁₂-C₁₃ etc.) of retinal exhibit distinct closely spaced peaks which overlap in that region. In addition, C-C-H in-plane rock modes are also sometimes coupled with the C-C stretch which makes the spectral evolution of this finger print modes even more challenging to rationalize. However, the most intense peak below 1200 cm⁻¹, which is known to be mostly uncoupled from C-C-H rock modes, follows a trend which can also be accounted for the change in the effective π -conjugation length. It has been also observed for linear conjugated polyenes^{80, 81} that the small reduction in effective conjugation length causes a blue shift of the mode below 1200 cm⁻¹ but further reduction of effective conjugation length eventually results in red shift. This matches very well the frequency shift of AT ASR (Fig. 9), where it first undergoes an initial blue shift until 100 fs and subsequently red shifts until 200 fs. This is the delay which has been interpreted above as the time to reach the 90° twisted state for AT ASR. Afterwards, the effective conjugation length is expected to increase again due to the slow formation of relatively planar K_{AT}. Thus, it causes a blue shift of the finger print mode. This effect is much weaker for 13C (~LA) isomer which results in no observable shift. This once again hints at a smaller rotation around C₁₃=C₁₄ in 13C isomer compared to AT isomer in ASR.

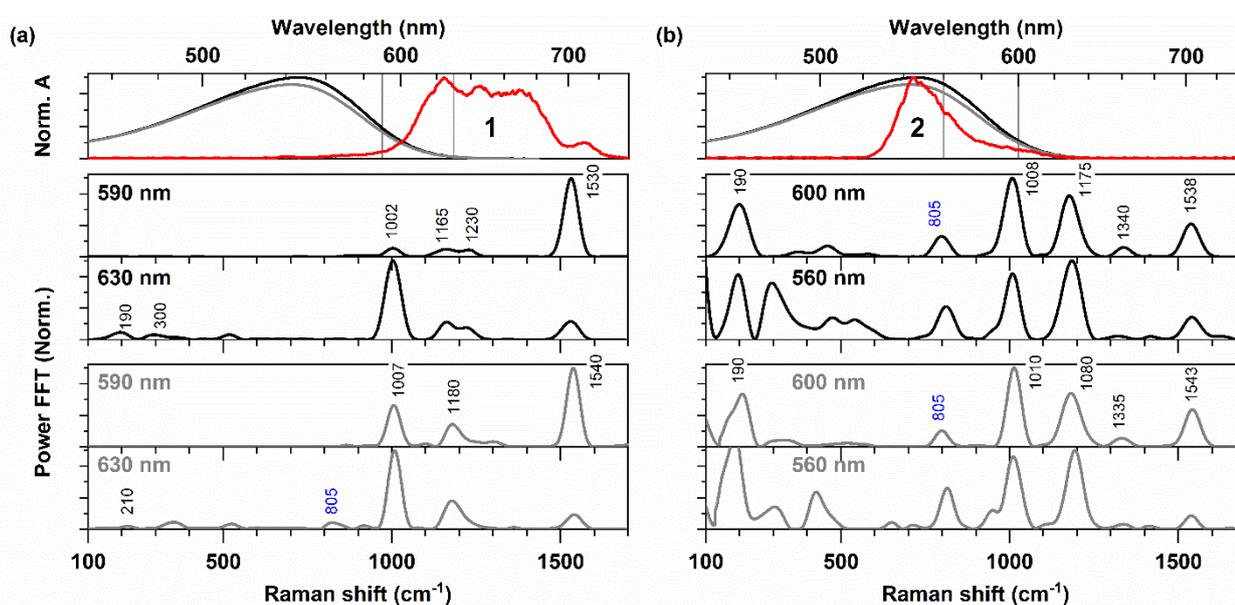


Figure 12: DFWM FFT spectra obtained by using near-resonant (1) and resonant (2) excitation spectra for DA (black) and LA (grey) ASR detected at different probing wavelengths shown by vertical grey line in (a) and (b).

In the context of frequency-shift for the excited state modes it is also important to note that the experimental time resolution cannot resolve the very fast red-shift taking place as a result of the BLA. This red-shift must take place according to numerical calculations in time-scales around 20 fs.³² The fact that an initial blue-shift is observed in all measurements at delays where BLA is temporally overlapping, possibly means that the *pure* blue-shift is potentially much bigger than the observed one.

Excited state evolution of low frequency (<400 cm⁻¹) and HOOP modes

A central result of pump-DFWM measurements was the observation regarding strong activity of the low frequency modes around 190 and 300 cm⁻¹ at positive actinic pulse delay which are absent in the GS ($T < 0$). Very similar low frequency peaks (160, 210 and 300 cm⁻¹) have also been observed previously for ASR in the transient absorption experiments⁵⁷ by probing in the near infrared region (950-1450 nm) which is far away from the ground state absorption and only covered by the stimulated emission. Furthermore, few strong Raman modes below 400 cm⁻¹ have been observed in the excited state of RPSB in solution⁵² by Kraack *et al.* and also observed more recently for rhodopsin²⁸ and Chanello-rhodopsin⁵⁸ by Schnedermann *et al.* Moreover, most important, the low frequency modes (e.g. ~190 cm⁻¹) in the pump-DFWM transients show faster dephasing than high frequency modes (e.g. C=C stretch) (Fig. S10). All these observations together clearly suggest that the low frequency Raman activities are originated from the ES manifold.

Another central result is the lack of HOOP activity in the excited state in our pump-DFWM (and pump-IVS) measurements which is surprising. There are two plausible explanations for this observation. The surrounding protein environment in ASR, which is different compared to rhodopsin and BR, where the HOOP activity is stronger,^{23, 28} can significantly reduce the Raman transition probability of a specific mode by modifying the retinal pocket. Secondly, the activity of the HOOP mode of BR has been observed to be strongly dependent on the center wavelength of DFWM spectra, used to create the vibrational coherence.⁵⁰ The Raman activity of this mode was found to be mainly enhanced for blue detuned excitation. In contrast, DFWM excitation spectra in our study, were red detuned compared to the ground state absorption (Fig. 2).

To clarify the lack of HOOP activity as well as to find out the activation mechanism of the low frequency (<400 cm⁻¹) excited state modes, we carried out two additional set of DFWM measurements with two different excitation spectra (Fig. 12). Spectrum 1 is the same as used before (Fig. 2) which is almost non-resonant to GS absorption and hence, only capable of exciting GS modes and also possibly the ES modes below 500 cm⁻¹ due to small overlap with GS. Spectrum 2, however, being completely resonant to the GS absorption, can directly excite all GS as well as ES vibrational modes. Two major changes (Fig. 12) were observed for both DA and LA ASR by tuning the DFWM

spectrum, namely (i) amplification of low-frequency and HOOP modes and the (ii) frequency-shift of high-frequency modes.

(i) Relative amplification of low frequency (<400 cm⁻¹) and HOOP modes: The FFT spectra for non-resonant spectrum (Fig. 12(a)), detected at 590 and 630 nm, show the activity mostly above 1000 cm⁻¹ along with a few weak low frequency modes (~190, 300 and 510 cm⁻¹), as observed before (Fig. 4). In contrast, a strong relative amplification of the FFT amplitude of low frequency modes (<400 cm⁻¹) with respect to high frequency modes (>1000 cm⁻¹), was observed for the FFT spectra in the resonant DFWM experiment (Fig. 12(b)) compared to the non-resonant DFWM measurement (Fig. 12(a)). On the one hand, these low frequency (<400 cm⁻¹) modes have been observed to be completely absent in the non-resonant steady state Raman spectrum (Fig. 4(i)-(j)) and therefore, these modes have been attributed to ES manifold. This assignment nicely matches the weak Raman activity of these low frequency modes in the *non-resonant* DFWM/IVS FFT spectrum (Fig. 12(a)). On the other hand, the same set of low frequency (<400 cm⁻¹) modes is *relatively* enhanced in resonant DFWM measurements, in comparison to the high frequency (>1000 cm⁻¹) vibrational modes. One obvious explanation for this observation would be that the low frequency modes are much more Franck-Condon active than the high-frequency modes. However, an alternative interpretation can also be drawn if one considers different activation mechanisms for the high and low frequency modes as it was reported in theoretical studies on retinal chromophore model³⁰ as well as experimental studies for BR⁵⁰ and RPSB in solution.^{52, 53} In these studies, the high frequency (>1000 cm⁻¹) modes were reported to be Frank-Condon active, whereas the low frequency (<400 cm⁻¹) torsional modes were considered to be impulsively excited by the high frequency modes via internal vibrational energy redistribution (IVR). This is further supported by our observation of the delayed rise of low frequency amplitude compared to the high frequency modes in transient pump-DFWM spectra (Fig. S8).

The second question remain about the reason behind the passiveness of the HOOP mode Raman activity in our pump-DFWM experiments. DFWM experiments with two different spectra (Fig. 12) show that the mode at 805 cm⁻¹, previously assigned as HOOP mode, shows a significantly strong amplitude in resonant DFWM experiment (Fig. 12(a)) which was very weak in the GS (Fig. 12(b)). This shows that the HOOP modes are induced only when it is probed with blue-detuned DFWM spectra not with red-detuned DFWM spectrum, used in our pump-DFWM experiment. This observation is very similar to BR⁵⁰ and thus seems to be an intrinsic molecular property of retinal.

(ii) Frequency shift of the high frequency (>1000 cm⁻¹) modes: The high frequency mode e.g. C=C stretch (1530 cm⁻¹ for DA and 1540 cm⁻¹ for LA) in non-resonant DFWM FFT spectrum (Fig. 12(a)) gets blue shifted in resonant DFWM FFT spectrum (Fig. 12(b)) (1538 cm⁻¹ for DA and 1543 cm⁻¹ for LA). A similar trend is also observed for CH₃ rocking modes: 1002 cm⁻¹ shifts

to 1008 cm^{-1} for DA while 1007 shifts to 1010 cm^{-1} for LA. This is expected since non-resonant DFWM FFT spectra contain only GS modes whereas resonant DFWM FFT spectra contain the mixture of GS and ES modes in the high frequency region. As we have shown above, both C=C stretch and CH_3 rock modes show a frequency blue shift in the ES compared to the GS (Fig. 8).

Pre-twisting and isomerization dynamics

In the context of retinal photo-chemistry, pre-twisting has been claimed to be one of the potential elements that can accelerate the isomerization process. In particular, the pre-straining inside the protein pocket has been repeatedly shown for visual rhodopsin by both X-Ray crystallography,⁹ NMR,⁴⁰ resonant Raman studies^{44, 69, 71, 77} and theoretical simulations^{42, 82} which accounts for its fast ballistic IC dynamics. Although non-resonant Raman spectra (Fig. 4), as reported in this study, speak for a relatively more pre-twisted geometry of 13C isomer in ASR, no significant difference in distortion around the $\text{C}_{13}=\text{C}_{14}$ between AT and 13C isomers has been pointed out by the X-ray crystallographic study of ASR with 2.0 Å resolution.⁸³ Very similar observations were made for BR where the 13C isomer shows 3 times faster kinetics⁸⁴ compared to AT but the structural data report both the isomers to be nearly planar without any indication of protein-catalyzed strain like visual rhodopsin. However, an indirect evidence of weak pre-straining for 13C isomer was found in the NMR⁷⁸ and resonance Raman⁷⁰ studies of BR although it seems to be negligible compared to 11-cis isomer of visual rhodopsin. More recently, a closer inspection on the X-ray data depicted a twist around $\text{C}_{14}-\text{C}_{15}$ bond in 13C isomer but not in AT isomer of BR.¹⁰ Similarly our classical GS dynamics simulation (Table S1) shows no difference in the distortion around $\text{C}_{13}=\text{C}_{14}$ bond for the AT and 13C isomers; rather it indicates the presence of a twist ($\sim 10^\circ$) around $\text{C}_{14}-\text{C}_{15}$ bond for 13C isomer whereas for AT isomer appears to be almost planar. Thus, this twist around the $\text{C}_{14}-\text{C}_{15}$ bond, which is adjacent to active $\text{C}_{13}=\text{C}_{14}$, may play a central role in the faster dynamics of 13C in ASR as claimed for 11C in visual rhodopsin.⁸⁵

This difference in distortions must originate from differences in the packing of the RPSB inside the retinal pocket. It fits the observation of faster dephasing (Fig. S9) of GS coherence of AT than 13C isomer which can be interpreted as stronger coupling of AT isomer to the surrounding than 13C. A recent femtosecond X-ray study²⁹ of BR has depicted the importance of the specific electrostatic interactions between protein and RPSB to guide the isomerization in certain direction. Hence, this difference in electrostatic interaction between AT and 13C isomers with the protein surrounding may lead to different trajectories during the reaction. It corroborates a FTIR study⁷² which showed that the sub-ps isomerization causes a stronger disruption of the H-bond between the surrounding water molecule and the protonated N-atom of the Schiff base in case of the AT isomer compared to 13C. This H-bond has been previously⁴⁶ suspected to be responsible for hindering the rotation of the protonated Schiff base and thus slowing down the isomerization around $\text{C}_{13}=\text{C}_{14}$ bond. This also matches our

observations for AT ASR of a slower frequency shift of fingerprint modes (C=C and C-C stretches, CH_3 rock) which are sensitive to localized structural changes and have been interpreted as the delayed formation of the twisted transition state for AT compared to 13C isomer.

Conclusion

This work has investigated the mechanistic origin of the huge dynamical differences observed in the isomerization of AT and 13C isomers in Anabaena Sensory Rhodopsin. In this regard, the evolution of the GS as well as of the ES structural changes of each isomer has been followed by applying DFWM, IVS, pump-DFWM and pump-IVS spectroscopy techniques. The present experiments were able to unveil three major structural and dynamical differences in the isomerization of each isomer: (i) HOOP activity in the GS is stronger for the 13C ASR than for AT ASR, (ii) large (up to 20 cm^{-1}) and delayed transient frequency blue shifts observed for the C=C stretching and CH_3 rock modes in the excited state of AT ASR and (iii) delayed Raman activity increase of low frequency modes ($< 400 \text{ cm}^{-1}$).

These experimental findings depict very different isomerization scenarios for each isomer. The stronger HOOP activity at about 805 cm^{-1} in the GS spectra of LA ASR indicates that 13C isomer is already more pre-twisted in the GS than the AT ASR isomer inside the retinal pocket. Bearing analogy to BR and supported by theoretical calculations, we have assigned this mode to $\text{C}_{14}-\text{H}$ wag. Our results point to a distortion located around the $\text{C}_{14}-\text{C}_{15}$ bond of the 13C isomer, which is neighboring to the isomerizing $\text{C}_{13}=\text{C}_{14}$ bond. The evolution of frequency shifts of high frequency modes, in particular of the C=C stretching and CH_3 rock modes, depicts a much slower formation of the twisted configuration for AT ASR compared to 13C ASR. Finally, the delayed rise of amplitude in the transient nonlinear Raman spectra as well as the stronger relative amplification of the low frequency modes from non-resonant to resonant DFWM experiments indicates that these modes are potentially activated by the C=C bond via IVR mechanism similarly to previous observations in BR.⁵⁰

These discoveries have profound implications in understanding the mechanism of the primary events in retinal proteins. 13C ASR and AT ASR show several dynamic and spectral features known for other respective isomers in other retinal proteins, in particular BR. Perhaps a major point is the very distinct evolution of high frequency modes in the excited states of 13C and AT ASR. While e.g. C=C stretching and CH_3 modes are not reactive coordinates *per se*, they certainly reflect the structural changes taking place at localized positions along the retinal during the isomerization. The delayed decrease of the conjugated double bond length observed for AT ASR in the excited state compared to 13C ASR surely follows the previous proposal about the presence of barrier in the excited state manifold of AT ASR. Nevertheless, our findings suggest that the longer dynamics observed for AT ASR may originate due to more than one factor, namely a barrier in the excited state and the lack of a pre-distortion (compared to 13C) in the ground state. And finally, although the HOOP Raman activity in 13C ASR is

much lower than for 11Cis in visual rhodopsin, the excited state lifetimes are not very different. This further reinforces that not a single effect is playing a role in determining the excited state lifetime, as has been advocated in the past. We expect that a time resolved vibrational spectroscopy study of specific point mutations of ASR will be able to pinpoint the role of barriers and pre-distortions and unveil the extent of this effect.

Conflicts of interest

There are no conflicts to declare.

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